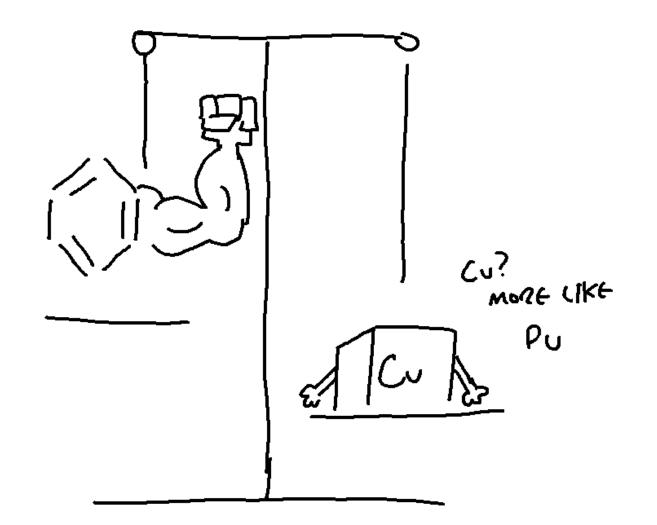
# Stable Dispersion of Graphene Synthesis

Taj Johnson and Arjun Nepal

## Hexagons are the Bestagons Graphene CARBON Graphite Carbon nanotube $C_{60}$

Dinadayalane, T.C., Leszczynski, J. Handbook of Computational Chemistry (2012)

#### What does it do? A lot.



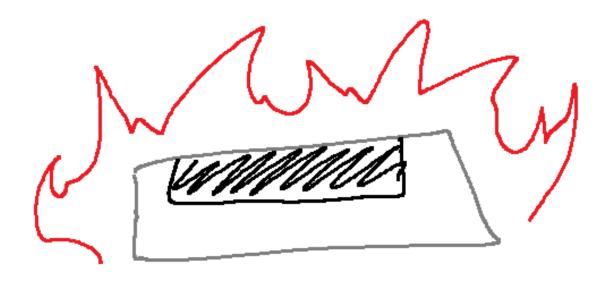
- Durable
- Light
- VERY thin
- Flexible
- Conductive
- Thermal Conductivity
- Magnetism
- Filtration
- Lubrication
- Waterproof
- And more

#### Wow, how do I get it?

Top Down

Bottom up



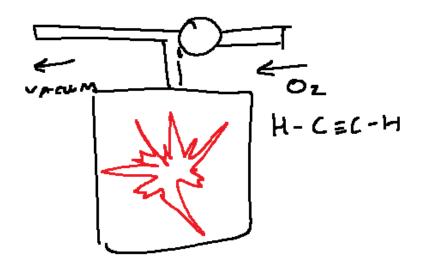




#### The Easy Way

#### Controlled Hydrocarbon Detonation

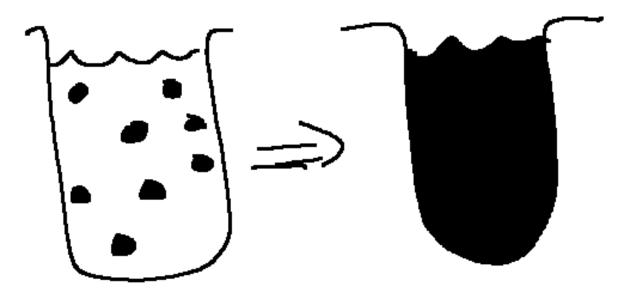
 Oxygen + Acetylene = Graphene Dust + Water Vapor

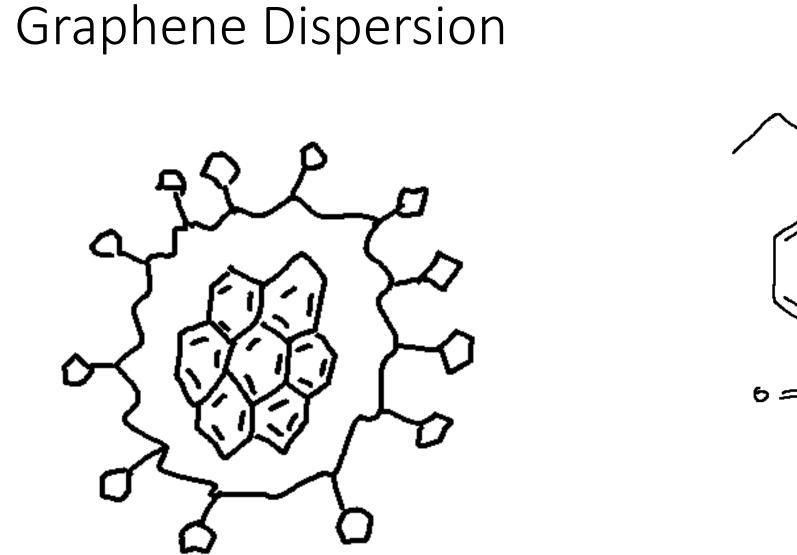




#### My Experiment

- Highest Concentration of Graphene
- Lowest Concentration of PSS





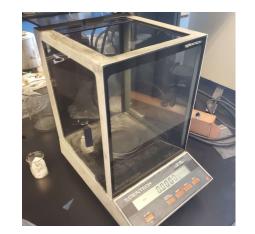
6=5=0 0

Polystyrene Sulfonate (PSS)

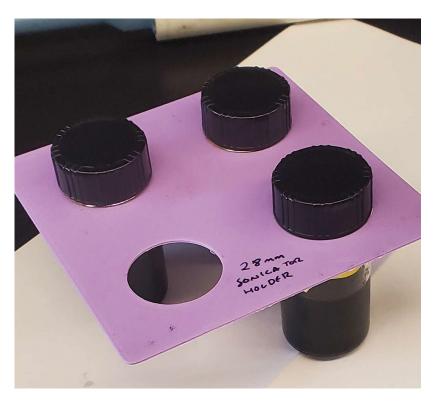
#### Procedure

- Measure
- Shake
- Wait
- Repeat

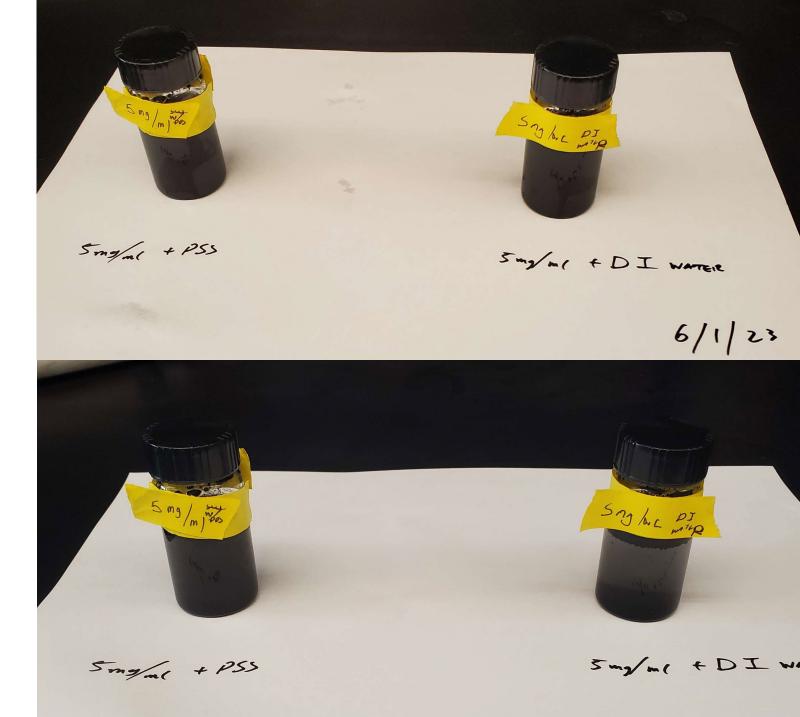








#### Experiment #1

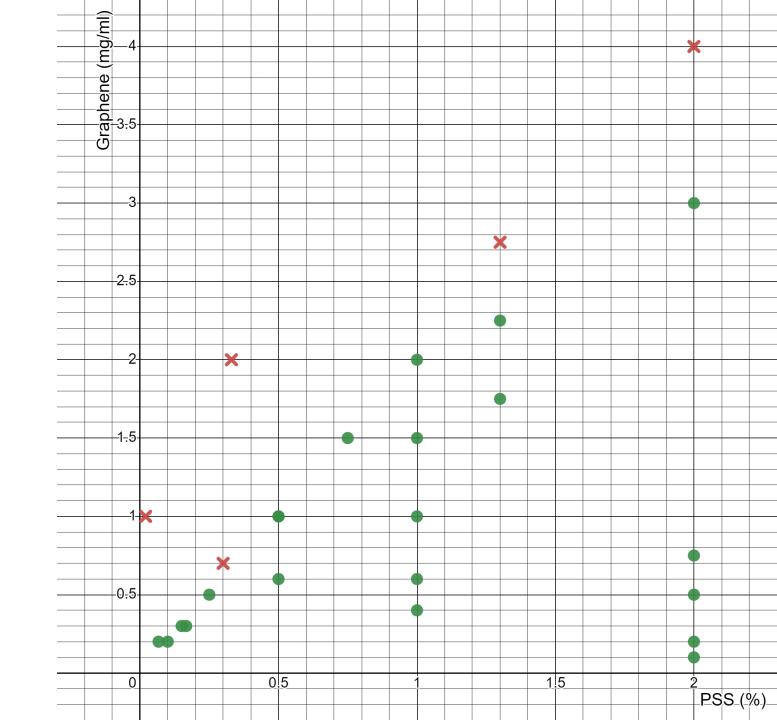


#### Experiment #2 - 14



### All Data Plot

- Green: Dispersed
- Red: Settlement



#### Results

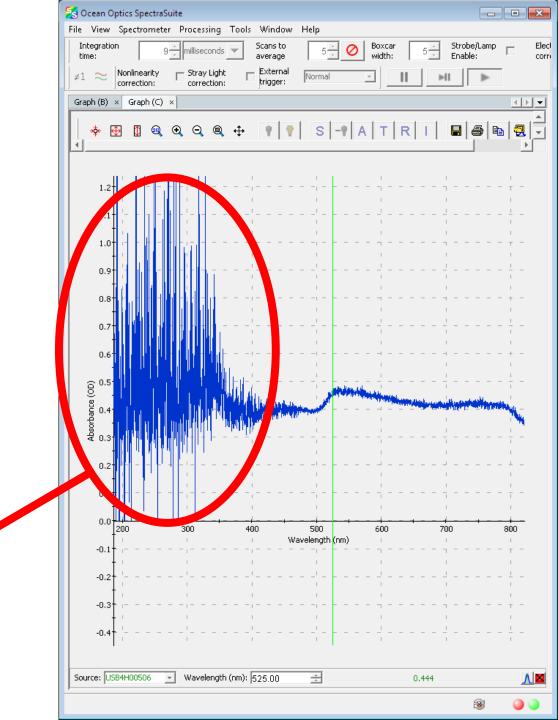
- PSS : Graphene Ratio
  - 1% PSS per 2mg/ml



#### Further Research

- Maximum saturation
- More Precision
- Testing Physical Properties
- Using UV-Vis

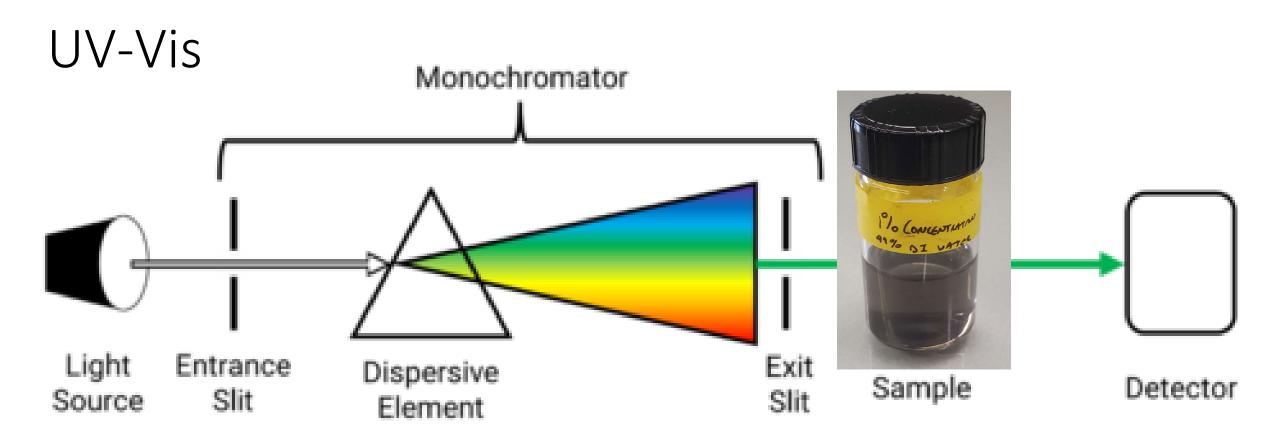
#### **Too Much Interference**



#### **Concentration Calculation Beer-Lambert**

$$A = \varepsilon cl$$
$$A_1 = \varepsilon c_1 l$$
$$A_2 = \varepsilon c_2 l$$
$$\frac{A_1}{A_2} = \frac{c_1}{c_2}$$

- *A* = Absorbance
- $\varepsilon$  = Molar Absorption Coefficient
- *c* = Molar Concentration
- *l* = Optical Path Length



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#### Questions?

